Noms.	Formules.
Sulphas ammonico-magnesicus c. aquâ .	$2\dot{N}H^6\ddot{S}Aq^2+\ddot{M}g\ddot{S}^2Aq^{10}$
calcico-natricus (Glauberit) .	$\ddot{N}a\ddot{S}^2+\ddot{C}a\ddot{S}^2$
cuprico-kalicus	$\ddot{K}\ddot{S}^2+\ddot{C}u\ddot{S}^2$
cum aquâ . .	$\ddot{K}\ddot{S}^2Aq^2+\ddot{C}u\ddot{S}^2Aq^{10}$
ferrico-kalicus	$\ddot{K}\ddot{S}^2+2\ddot{F}e\ddot{S}^3$
Tartras kalico-ferrosus	$\ddot{K}\bar{T}^2+\ddot{F}e\bar{T}^2$
kalico-natricus	$\ddot{K}\bar{T}^2+\ddot{N}a\bar{T}^2$
cum aquâ . .	$\ddot{K}\bar{T}^2+\ddot{N}a\bar{T}^2Aq^{20}$?
stibicus	$3\ddot{K}\bar{T}^2Aq^2+4\dot{S}b\bar{T}^3Aq^3$?
Tritartroboras kalicus (cremor tartari solubilis)	$\ddot{K}\ddot{B}^6+2\ddot{K}\bar{T}^6$
natricus	$\ddot{N}a\ddot{B}^6+2\ddot{N}a\bar{T}^6$
Wolframias ferroso-manganosus . . . . (wolfram)	$\ddot{M}n\ddot{W}^2+3\ddot{F}e\ddot{W}^2$